



Recent Field Applications of Activated Persulfate Including Petroleum Sites

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Presentation Outline

- **Brief Introduction to Persulfate Chemistry**
- **Activated Persulfate Treatment of:**
 - **BTEX**
 - **GRO**
 - **#6 Fuel Oil NAPL**
 - **MTBE**

Acknowledgements

Scott Pittenger



Larry Kinsman



Jason Swearingen



Isaac Aboulafia



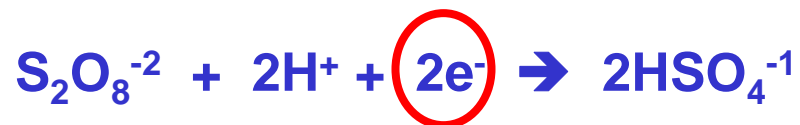
John Haselow, PhD



Introduction to Persulfate Chemistry

Persulfate Oxidation Principles:

Persulfate Anion: $E^0 = 2.12 \text{ v}$



+ Activation

Sulfate-free radical:



- Activation Produces a Radical Which is More Powerful and Kinetically Fast
- Proper Activation Method is Based on Contaminant, Site Lithology, and Hydrogeology

Introduction to Persulfate Chemistry

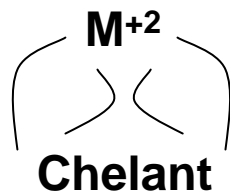
Four Primary Methods to Activate Klozur[®] Persulfate

1. Heat



- kinetically fast
- capable of destroying wide range of contaminants
- 35 – 45 C

2. Metal Chelates



- slower kinetics; longer lived
- capable of destroying chlorinated ethenes, BTEX, PAHs, MTBE
- target minimum 150 ppm F

3. Hydrogen Peroxide



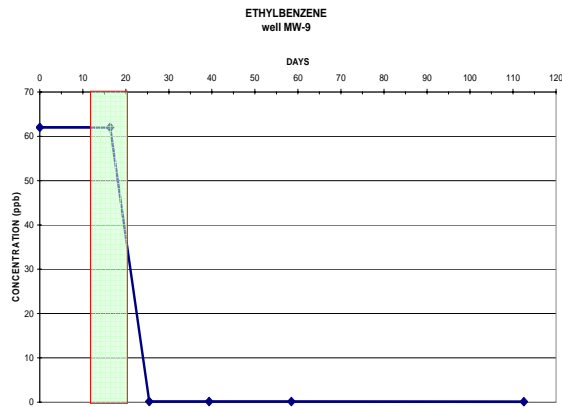
- kinetically fast
- capable of destroying wide range of organics
- typical concentration: 5 moles peroxide / mole persulfate

4. High pH

- kinetically fast
- capable of destroying wide range of organics
- pH > 10.5

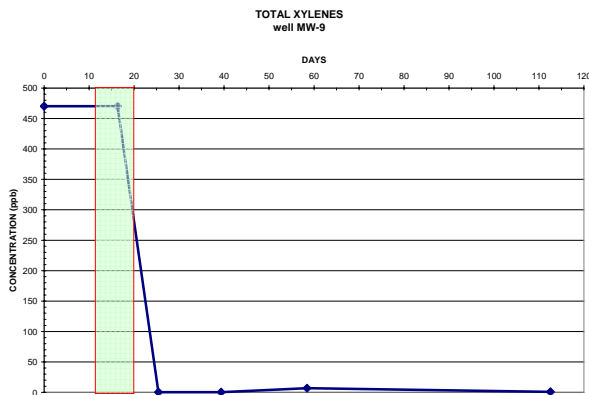
Ethylbenzene & Xylenes; chelated iron activation

Site: former maintenance yard in Florida
Contaminants: ethylbenzene (130 ppb) and xylenes (1000 ppb)
History: one year of bioremediation prior to ISCO



Application

1800 gal of 20% Klozur® soln
 297 gal of 2.5% Fe-citrate soln
 Injected over 3 days



Conclusions

GW contaminant concentration dropped below LMDLs within 7 days. Klozur® persulfate activated by Fe(II)-citrate effectively treated ethylbenzene and xylenes to meet the clean-up goals for this site with a single injection event targeting a limited impacted area.

Pipeline Release - BTEX; Calcium Peroxide activation



Site Information:

- former pipeline release
- ave benzene in groundwater is **4.7 ppm**

Data courtesy of:



Calcium Peroxide activation:

- high pH activation (> 11)
- peroxide activation Calcium peroxide generates hydrogen peroxide
- subsequent oxygen release over 9 months supports aerobic bioremediation
- generated sulfate supports long term anaerobic bioremediation

Field Implementation:

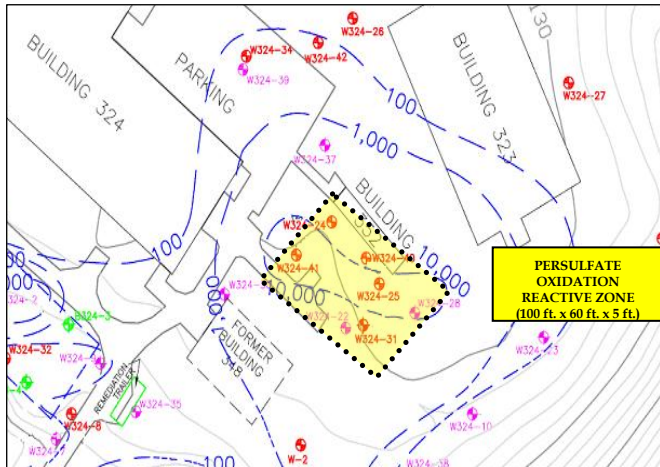
- 60,000 ft³ plume treated
- 25 direct push points: each received 50 gallons of Klozur persulfate + 100 gallons of PermeOx Plus
- 5 days of application time

Results:

- 4 months after application, benzene dropped from 4700 ppb to 3 ppb
- estimated cost savings over dig / haul and other chemistries: \$28,000 - \$50,000

GRO former tank farm; Hydrogen Peroxide activation

Data courtesy of



Site Information:

- Fort Belvoir, VA.
- former fuel tank farm with GRO, DRO
- silty sands with cobble and gravel deposits

Field Implementation:

- 5,225 lbs of Klozur persulfate
- 1000 lb of 17.5% peroxide
- addition over a 5 day period

TOTAL BTEX BASELINE CONCENTRATION ISOPLETH MAP WITH PERSULFATE OXIDATION REACTIVE ZONE.



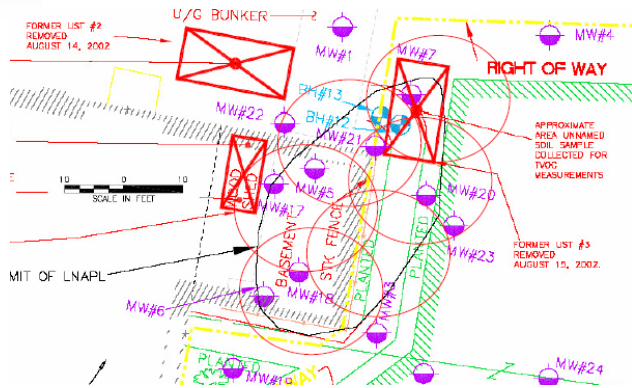
	Sample Date	Benzene	Toluene	Ethylbenzene	Xylenes	Total BTEX
W324-22	9/27/2007	<2.0	<2.0	3.5	4.9	8.4
	9/7/2007	<1.0	<1.0	5.9	1.0	6.9
	<i>Persulfate Oxidation Event (8-26-07)</i>					
	4/2/1999	7,820	6,080	1,930	8,260	24,090
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Example of results from one well

Results:

- 1 month results indicate 88 – 99% reduction in total BTEX
- larger scale injection planned

#6 Fuel Oil NAPL; hydrogen peroxide activation



Site Information:

- historic building in MA
- leaking UST
- ave TPH in ground is 13,750 mg / kg
- NAPL thickness: 1 – 30 inches
- total NAPL: 2000 gallons
- target clean-up goal: < 1/2 NAPL on site

Data courtesy of



Field Implementation:

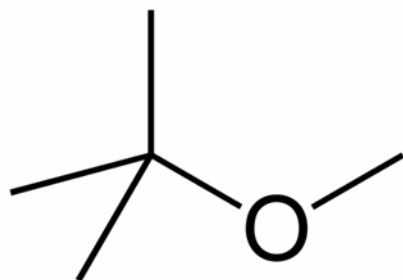
- 2 weeks of hydrogen peroxide activated persulfate application
- sub-surface vapor extraction system



Parameter	Pre-ISCO (Baseline)	Post-ISCO Value
Estimated NAPL volume	2,000 gallons	600 gallons
Average NAPL thickness	1 to 30 inches	< 1/2-inch
Average TPH concentration in soils	13,750 mg/Kg	8,500 mg/KG

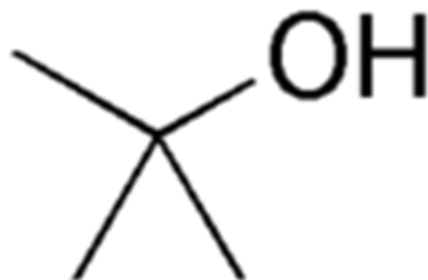
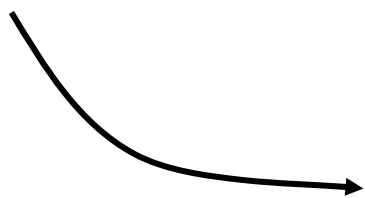
Results a couple of weeks after implementation: site closure with deed restrictions

MTBE

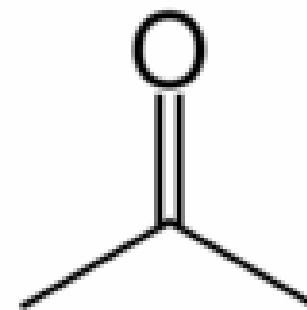
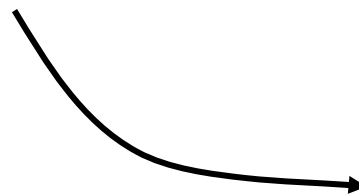


Structure of MTBE

- gasoline additive to enhance fuel burning
- over 20% of US drinking water supply contaminated by MTBE

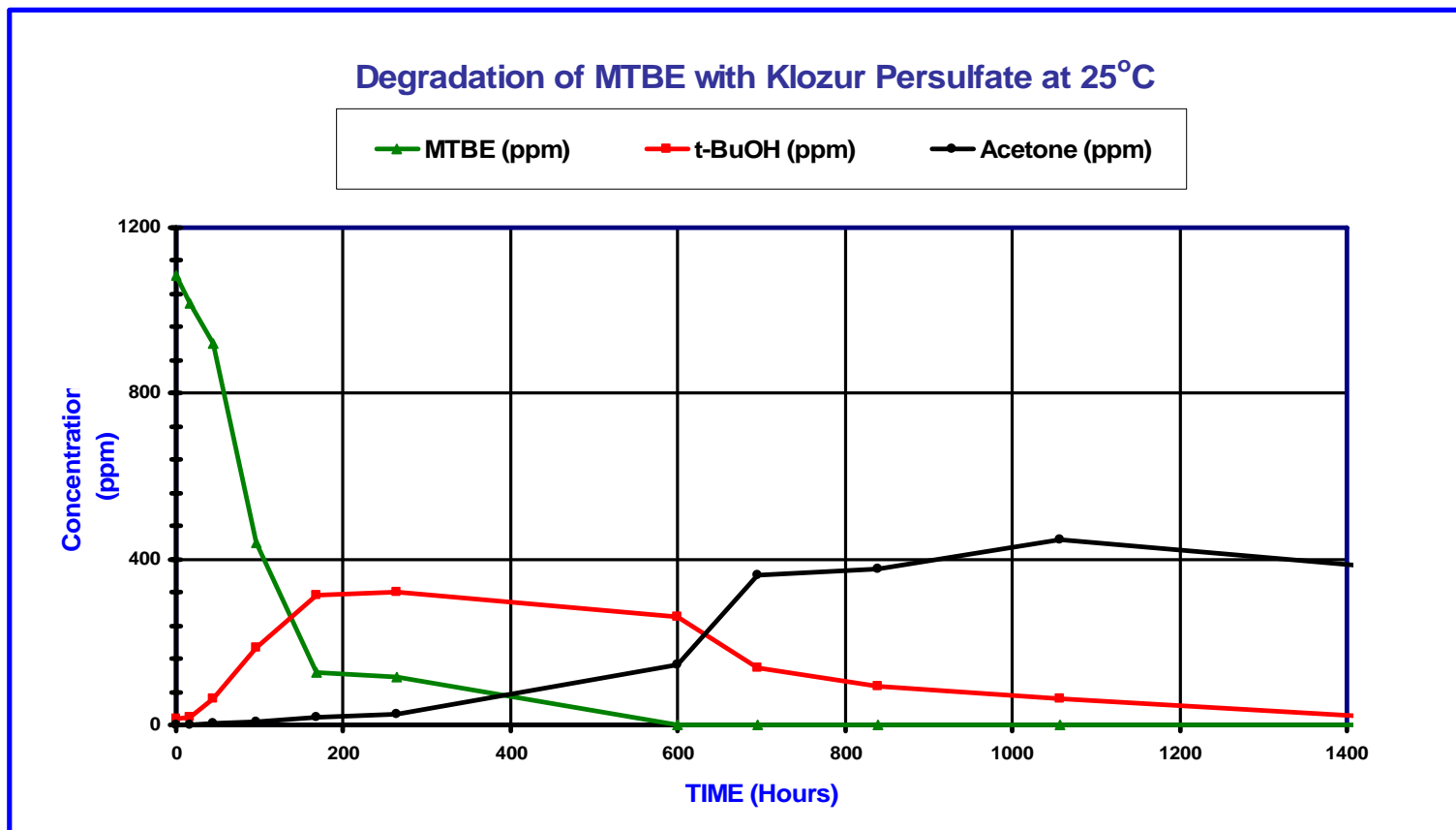


Structure of tert-butyl alcohol



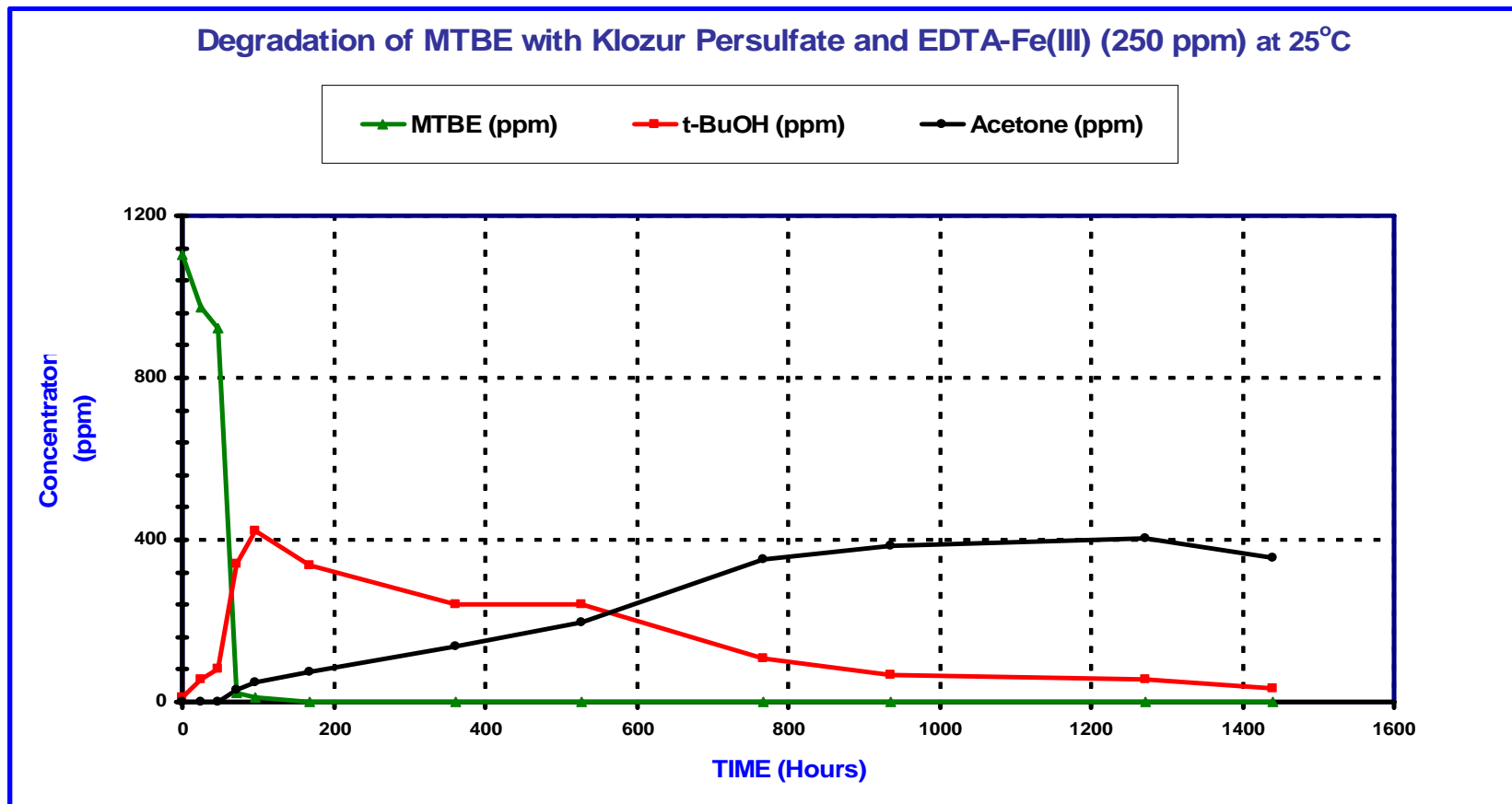
Structure of acetone

MTBE



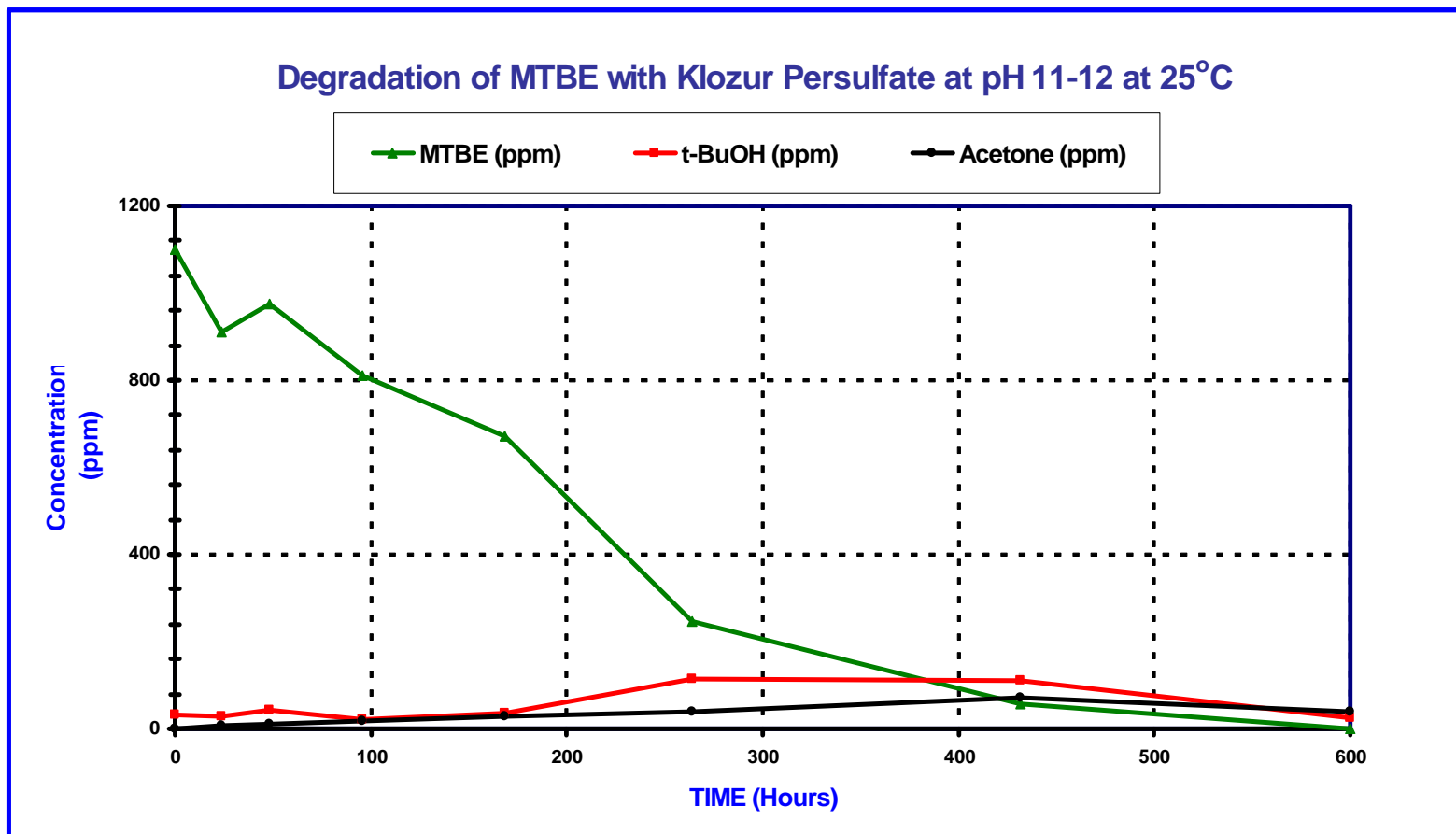
- un-activated persulfate oxidation of MTBE can lead to significant formation of TBA and acetone

MTBE



- FeEDTA persulfate rapidly destroys MTBE, but has much slower kinetics on TBA and acetone

MTBE



- High pH activated persulfate oxidation of MTBE does not generate significant quantities of TBA and acetone.

Benzene / Naphthalene / MTBE; heat / high pH activation

Petroleum Hydrocarbons; Heat / Alkaline Activation

Redox Tech, LLC

Location	Contaminant (ug/L)	Pre-Injection Concentration (ug/L)	Post-Injection Concentration	% Reduction
Blackstone, VA	Benzene	1,600	78	95.1%
Blackstone, VA	MTBE	1,300	360	72.3%
Clayton, DE	Benzene	519	7.4	98.5%
Clayton, DE	MTBE	16,100	233	98.5%

Location	Contaminant (ug/L)	Pre-Injection Concentration (ug/L)	Post-Injection Concentration	% Reduction
Hagaman, NY	Xylene	2,778	22	99.2%
Lexington, NC	Naphthalene	> 1,000,000	< 1,000	99.9%

Summary

- Activated persulfate is effective at treating a range of TPH related compounds
 - BTEX
 - PAHs (ex: naphthalene, chrysene, anthracene, pyrene, etc)
 - GRO
 - DRO
 - ORO
 - Long chain alkanes
 - Fuel oxygenates: MTBE, TBA
- The choice of activation method is dependent on target contaminant and site lithology / hydrogeology
- Activated persulfate can also effectively treat chlorinated solvents (ex: PCE, TCE, TCA, methylene chloride)

Questions ?

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